

catena-Poly[triethylammonium [[triphenyltin(IV)]- μ -3,3'-dihydroxy-4,4'-methylenedi-2-naphthoate]]

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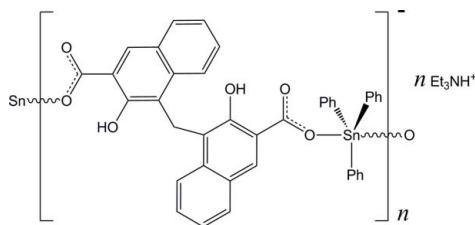
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Key indicators: single-crystal X-ray study; $T = 298$ K; mean $\sigma(\text{C}-\text{C}) = 0.018$ Å; R factor = 0.070; wR factor = 0.182; data-to-parameter ratio = 12.1.

The title compound, $\{(\text{C}_6\text{H}_{16}\text{N})[\text{Sn}(\text{C}_6\text{H}_5)_3(\text{C}_{23}\text{H}_{14}\text{O}_6)]\}_n$, has an infinite chain structure, formed through monodentate carboxylate groups of the pamoic acid anion. The anion bridges two symmetry-related Sn(IV) ions and the resulting polymeric chains are parallel to [201] in the crystal. Et_3NH^+ cations are inserted between the chains. The coordination of the Sn(IV) atom is completed by three phenyl ligands, giving a distorted trigonal-bipyramidal geometry.

Related literature

For related polymeric organotin structures, see: Ma *et al.* (2008).



Experimental

Crystal data

 $(\text{C}_6\text{H}_{16}\text{N})[\text{Sn}(\text{C}_6\text{H}_5)_3(\text{C}_{23}\text{H}_{14}\text{O}_6)]$ $M_r = 838.53$

Monoclinic, Cc
 $a = 13.2590$ (14) Å
 $b = 16.3231$ (16) Å
 $c = 19.166$ (2) Å
 $\beta = 98.580$ (2)°
 $V = 4101.6$ (7) Å³

$Z = 4$
Mo $K\alpha$ radiation
 $\mu = 0.67$ mm⁻¹
 $T = 298$ K
 $0.24 \times 0.14 \times 0.11$ mm

Data collection

Siemens SMART CCD 1000 area-detector diffractometer
Absorption correction: multi-scan (*SADABS*; Siemens, 1996)
 $T_{\min} = 0.855$, $T_{\max} = 0.930$

10588 measured reflections
6019 independent reflections
4406 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.115$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.070$
 $wR(F^2) = 0.182$
 $S = 0.98$
6019 reflections
499 parameters
2 restraints

H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 1.53$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.84$ e Å⁻³
Absolute structure: Flack (1983),
2394 Friedel pairs
Flack parameter: -0.04 (4)

Table 1

Selected geometric parameters (Å, °).

Sn1—C36	2.154 (15)	Sn1—O4	2.227 (12)
Sn1—C24	2.157 (14)	Sn1—O1	2.314 (12)
Sn1—C30	2.180 (11)		
C36—Sn1—C24	110.7 (4)	C30—Sn1—O4	95.1 (4)
C36—Sn1—C30	139.3 (5)	C36—Sn1—O1	86.3 (5)
C24—Sn1—C30	109.5 (5)	C24—Sn1—O1	89.2 (5)
C36—Sn1—O4	91.1 (5)	C30—Sn1—O1	88.6 (4)
C24—Sn1—O4	89.3 (5)	O4—Sn1—O1	176.3 (4)

Data collection: *SMART* (Siemens, 1996); cell refinement: *SAINT* (Siemens, 1996); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXTL*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: BH2324).

References

- Flack, H. D. (1983). *Acta Cryst.* **A39**, 876–881.
Ma, C., Wang, Q. & Zhang, R. (2008). *Eur. J. Inorg. Chem.* pp. 1926–1934.
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supplementary materials

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catena-Poly[triethylammonium [[triphenyltin(IV)]- μ -3,3'-dihydroxy-4,4'-methylenedi-2-naphthoate]]

L. Liu and S. Gong

Comment

The title compound is a one-dimensional infinite chain structure linked by monodentate pamoic acid ligand, as shown in Figures 1 and 2. The geometry of tin atoms is distorted trigonal bipyramidal, surrounded axially by two oxygen atoms and equatorially by three carbon atoms of the phenyl groups. The axial angle O4—Sn1—O1 is 176.3 (4)°, close to linear arrangement. Three Sn-phenyl groups define the equatorial plane and the sum of the trigonal C—Sn—C angles is 359.5°, as expected for a bipyramidal geometry. The Sn1—O1 distance is 2.314 (12) Å, which is a bit longer than the covalent bond length Sn—O, but similar to those found in other reported triorganotin polymeric structures (*e.g.* Ma *et al.*, 2008).

Experimental

The reaction was carried out under nitrogen atmosphere. 4,4'-Methylenebis(3-hydroxy-2-naphthoic acid) (1 mmol) and triethylamine (2 mmol) were added to a stirred solution of benzene (30 ml) in a Schlenk flask and stirred for 0.5 h. Triphenyltin chloride (2 mmol) was then added and the reaction mixture was stirred for 12 h at 353 K. The resulting clear solution was evaporated under vacuum. The product was crystallized from dichloromethane to yield colourless blocks of the title complex (yield 83%. m.p. 458 K). Anal. Calcd (%) for C₄₇H₄₅N₁O₆Sn₁ (Mr = 838.53): C 67.32, H 5.41. Found (%): C 67.61, H 5.68.

Refinement

All H atoms were placed geometrically and treated as riding on their parent atoms with C—H = 0.93 (aromatic CH), C—H = 0.97 (methylene), C—H = 0.96 (methyl), N—H = 0.91, and O—H = 0.82 Å. Isotropic displacement parameters for H atoms were calculated as $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{carrier atom})$ or $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{carrier atom})$.

Figures

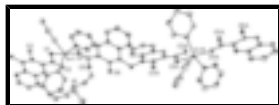


Fig. 1. The molecular structure of the compound, showing 30% probability displacement ellipsoids.

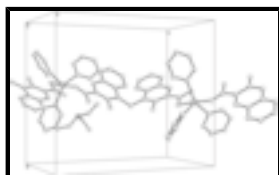


Fig. 2. The unit cell of the title compound.

supplementary materials

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Crystal data

(C ₆ H ₁₆ N)[Sn(C ₆ H ₅) ₃ (C ₂₃ H ₁₄ O ₆)]	$F(000) = 1728$
$M_r = 838.53$	$D_x = 1.358 \text{ Mg m}^{-3}$
Monoclinic, Cc	Melting point: 458 K
Hall symbol: C -2yc	Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$
$a = 13.2590 (14) \text{ \AA}$	Cell parameters from 2599 reflections
$b = 16.3231 (16) \text{ \AA}$	$\theta = 2.4\text{--}25.0^\circ$
$c = 19.166 (2) \text{ \AA}$	$\mu = 0.67 \text{ mm}^{-1}$
$\beta = 98.580 (2)^\circ$	$T = 298 \text{ K}$
$V = 4101.6 (7) \text{ \AA}^3$	Block, colourless
$Z = 4$	$0.24 \times 0.14 \times 0.11 \text{ mm}$

Data collection

Bruker SMART CCD 1000 area-detector diffractometer	6019 independent reflections
Radiation source: fine-focus sealed tube graphite	4406 reflections with $I > 2\sigma(I)$
φ and ω scans	$R_{\text{int}} = 0.115$
Absorption correction: multi-scan (SADABS; Siemens, 1996)	$\theta_{\text{max}} = 25.0^\circ$, $\theta_{\text{min}} = 2.0^\circ$
$T_{\text{min}} = 0.855$, $T_{\text{max}} = 0.930$	$h = -15 \rightarrow 13$
10588 measured reflections	$k = -15 \rightarrow 19$
	$l = -22 \rightarrow 22$

Refinement

Refinement on F^2	Secondary atom site location: difference Fourier map
Least-squares matrix: full	Hydrogen site location: inferred from neighbouring sites
$R[F^2 > 2\sigma(F^2)] = 0.070$	H-atom parameters constrained
$wR(F^2) = 0.182$	$w = 1/[\sigma^2(F_o^2) + (0.1004P)^2]$
$S = 0.98$	where $P = (F_o^2 + 2F_c^2)/3$
6019 reflections	$(\Delta/\sigma)_{\text{max}} = 0.001$
499 parameters	$\Delta\rho_{\text{max}} = 1.53 \text{ e \AA}^{-3}$
2 restraints	$\Delta\rho_{\text{min}} = -0.84 \text{ e \AA}^{-3}$
0 constraints	Absolute structure: Flack (1983), 2394 Friedel pairs
Primary atom site location: structure-invariant direct methods	Flack parameter: $-0.04 (4)$

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
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Sn1	0.76424 (5)	0.44960 (4)	0.08738 (4)	0.0419 (2)
N1	0.5704 (7)	0.6521 (6)	0.2187 (5)	0.059 (2)
H1	0.5506	0.6055	0.1943	0.071*
O1	0.5920 (9)	0.4307 (6)	0.0488 (6)	0.047 (2)
O2	0.5631 (6)	0.5121 (5)	0.1380 (4)	0.0522 (18)
O3	0.4525 (6)	0.4040 (5)	-0.0559 (4)	0.055 (2)
H3	0.5079	0.4003	-0.0305	0.082*
O4	0.9306 (9)	0.4593 (6)	0.1263 (6)	0.053 (3)
O5	0.9078 (7)	0.5519 (5)	0.2087 (5)	0.062 (2)
O6	1.0543 (6)	0.5914 (5)	0.3030 (4)	0.057 (2)
H6	0.9977	0.5932	0.2790	0.086*
C1	0.5324 (9)	0.4761 (7)	0.0813 (6)	0.046 (3)
C2	0.3574 (9)	0.5325 (6)	0.0775 (6)	0.046 (3)
H2	0.3789	0.5536	0.1223	0.055*
C3	0.4245 (8)	0.4854 (7)	0.0451 (6)	0.046 (3)
C4	0.3884 (14)	0.4520 (7)	-0.0235 (9)	0.044 (4)
C5	0.2893 (10)	0.4648 (8)	-0.0581 (7)	0.043 (3)
C6	0.2219 (8)	0.5155 (7)	-0.0229 (6)	0.045 (2)
C7	0.2572 (8)	0.5490 (6)	0.0438 (6)	0.043 (2)
C8	0.1909 (9)	0.6002 (7)	0.0782 (6)	0.054 (3)
H8	0.2134	0.6211	0.1230	0.065*
C9	0.0950 (9)	0.6178 (8)	0.0452 (6)	0.056 (3)
H9	0.0535	0.6536	0.0658	0.067*
C10	0.0601 (9)	0.5819 (8)	-0.0195 (7)	0.058 (3)
H10	-0.0069	0.5915	-0.0402	0.069*
C11	0.1191 (9)	0.5338 (7)	-0.0532 (6)	0.051 (3)
H11	0.0927	0.5120	-0.0969	0.061*
C12	0.9632 (9)	0.5017 (8)	0.1802 (6)	0.051 (3)
C13	1.1374 (14)	0.4399 (9)	0.1843 (10)	0.047 (4)
H13	1.1113	0.4120	0.1433	0.056*
C14	1.0735 (8)	0.4935 (7)	0.2120 (6)	0.045 (2)
C15	1.1138 (8)	0.5372 (6)	0.2741 (5)	0.041 (2)
C16	1.2141 (8)	0.5256 (6)	0.3074 (6)	0.043 (2)
C17	1.2795 (9)	0.4682 (7)	0.2773 (7)	0.043 (3)
C18	1.2383 (10)	0.4244 (8)	0.2135 (7)	0.048 (3)
C19	1.3028 (8)	0.3696 (7)	0.1845 (6)	0.053 (3)
H19	1.2765	0.3398	0.1445	0.063*
C20	1.4025 (9)	0.3586 (8)	0.2130 (6)	0.058 (3)
H20	1.4439	0.3232	0.1920	0.069*
C21	1.4406 (10)	0.4009 (7)	0.2732 (7)	0.058 (3)
H21	1.5079	0.3922	0.2935	0.070*
C22	1.3820 (9)	0.4556 (7)	0.3046 (6)	0.049 (3)
H22	1.4111	0.4845	0.3444	0.058*
C23	0.2537 (9)	0.4235 (8)	-0.1279 (6)	0.045 (3)
H23A	0.2001	0.3855	-0.1205	0.055*
H23B	0.3102	0.3910	-0.1394	0.055*
C24	0.7979 (12)	0.3698 (9)	0.0041 (8)	0.046 (3)
C25	0.7434 (11)	0.2978 (8)	-0.0158 (6)	0.062 (3)
H25	0.6855	0.2851	0.0041	0.075*

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C26	0.7759 (12)	0.2445 (8)	-0.0659 (7)	0.068 (4)
H26	0.7404	0.1961	-0.0778	0.082*
C27	0.8593 (13)	0.2632 (10)	-0.0969 (9)	0.070 (4)
H27	0.8778	0.2291	-0.1317	0.084*
C28	0.9164 (11)	0.3327 (9)	-0.0770 (6)	0.068 (4)
H28	0.9743	0.3443	-0.0973	0.082*
C29	0.8866 (9)	0.3854 (8)	-0.0263 (6)	0.059 (3)
H29	0.9259	0.4314	-0.0124	0.071*
C30	0.7482 (9)	0.5751 (7)	0.0484 (6)	0.049 (3)
C31	0.6617 (11)	0.5948 (9)	-0.0028 (7)	0.069 (4)
H31	0.6134	0.5553	-0.0190	0.082*
C32	0.6520 (12)	0.6747 (9)	-0.0276 (7)	0.078 (4)
H32	0.5945	0.6888	-0.0593	0.093*
C33	0.7264 (12)	0.7352 (9)	-0.0065 (7)	0.075 (4)
H33	0.7178	0.7888	-0.0226	0.090*
C34	0.8115 (12)	0.7121 (8)	0.0385 (7)	0.072 (4)
H34	0.8659	0.7483	0.0491	0.086*
C35	0.8164 (11)	0.6348 (8)	0.0680 (6)	0.069 (4)
H35	0.8701	0.6233	0.1037	0.083*
C36	0.7411 (13)	0.3798 (10)	0.1791 (8)	0.054 (4)
C37	0.7771 (11)	0.4021 (10)	0.2489 (6)	0.071 (4)
H37	0.8141	0.4503	0.2579	0.086*
C38	0.7585 (12)	0.3532 (10)	0.3055 (7)	0.079 (4)
H38	0.7840	0.3682	0.3516	0.095*
C39	0.7018 (14)	0.2820 (11)	0.2924 (9)	0.074 (5)
H39	0.6892	0.2494	0.3299	0.088*
C40	0.6642 (13)	0.2591 (9)	0.2248 (8)	0.078 (4)
H40	0.6236	0.2126	0.2163	0.094*
C41	0.6876 (11)	0.3068 (9)	0.1687 (6)	0.066 (3)
H41	0.6665	0.2889	0.1228	0.079*
C42	0.5879 (11)	0.7165 (9)	0.1637 (7)	0.072 (4)
H42A	0.6405	0.6972	0.1376	0.086*
H42B	0.6122	0.7666	0.1877	0.086*
C43	0.4911 (13)	0.7356 (10)	0.1115 (8)	0.091 (5)
H43A	0.4565	0.6854	0.0968	0.136*
H43B	0.5094	0.7634	0.0711	0.136*
H43C	0.4467	0.7699	0.1340	0.136*
C44	0.4862 (11)	0.6733 (9)	0.2603 (7)	0.073 (4)
H44A	0.4241	0.6826	0.2276	0.088*
H44B	0.4745	0.6266	0.2894	0.088*
C45	0.5061 (15)	0.7465 (11)	0.3066 (8)	0.104 (6)
H45A	0.5742	0.7437	0.3319	0.156*
H45B	0.4583	0.7477	0.3395	0.156*
H45C	0.4987	0.7952	0.2782	0.156*
C46	0.6704 (11)	0.6326 (9)	0.2641 (7)	0.073 (4)
H46A	0.7178	0.6124	0.2342	0.087*
H46B	0.6985	0.6828	0.2862	0.087*
C47	0.6621 (14)	0.5712 (10)	0.3200 (8)	0.091 (5)
H47A	0.6168	0.5912	0.3508	0.137*

H47B	0.7283	0.5617	0.3467	0.137*
H47C	0.6359	0.5208	0.2987	0.137*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Sn1	0.0373 (3)	0.0414 (3)	0.0450 (3)	0.0002 (5)	-0.0006 (2)	-0.0007 (5)
N1	0.062 (6)	0.053 (6)	0.061 (6)	0.003 (5)	0.003 (5)	-0.009 (5)
O1	0.041 (6)	0.043 (5)	0.057 (6)	0.004 (4)	0.005 (4)	0.001 (4)
O2	0.048 (4)	0.055 (5)	0.050 (4)	0.006 (4)	-0.004 (3)	-0.006 (4)
O3	0.039 (4)	0.058 (5)	0.064 (5)	0.012 (4)	-0.004 (3)	-0.011 (4)
O4	0.042 (6)	0.063 (6)	0.050 (6)	-0.003 (5)	-0.005 (4)	-0.008 (5)
O5	0.043 (5)	0.072 (6)	0.067 (5)	0.008 (4)	-0.006 (4)	-0.010 (5)
O6	0.048 (5)	0.061 (5)	0.059 (5)	0.015 (4)	-0.003 (4)	-0.008 (4)
C1	0.046 (7)	0.044 (6)	0.047 (6)	0.001 (5)	0.004 (5)	0.007 (5)
C2	0.044 (7)	0.045 (6)	0.049 (6)	-0.004 (5)	0.008 (5)	-0.004 (5)
C3	0.037 (6)	0.046 (6)	0.053 (6)	-0.003 (5)	0.003 (5)	0.004 (5)
C4	0.038 (9)	0.039 (8)	0.052 (8)	0.007 (5)	0.003 (6)	0.000 (5)
C5	0.038 (7)	0.041 (7)	0.051 (7)	-0.004 (5)	0.007 (6)	0.001 (6)
C6	0.037 (6)	0.042 (6)	0.056 (6)	0.004 (5)	0.008 (5)	0.002 (5)
C7	0.035 (6)	0.039 (6)	0.054 (6)	-0.001 (5)	0.006 (4)	-0.001 (5)
C8	0.052 (7)	0.048 (7)	0.063 (7)	-0.003 (6)	0.006 (6)	-0.006 (6)
C9	0.037 (6)	0.056 (7)	0.076 (8)	0.003 (5)	0.012 (6)	0.001 (6)
C10	0.039 (7)	0.060 (7)	0.073 (8)	0.001 (6)	0.005 (6)	0.007 (6)
C11	0.039 (7)	0.051 (7)	0.061 (7)	0.004 (5)	0.002 (5)	0.003 (5)
C12	0.046 (7)	0.056 (8)	0.050 (6)	0.004 (6)	0.001 (5)	0.006 (6)
C13	0.036 (9)	0.049 (8)	0.054 (9)	-0.001 (6)	0.004 (6)	-0.004 (6)
C14	0.038 (6)	0.041 (6)	0.053 (6)	-0.002 (5)	0.002 (5)	0.000 (5)
C15	0.039 (6)	0.039 (6)	0.043 (5)	0.002 (5)	-0.002 (4)	0.001 (4)
C16	0.037 (6)	0.038 (6)	0.053 (6)	-0.002 (5)	0.000 (5)	0.002 (5)
C17	0.038 (7)	0.043 (7)	0.048 (7)	0.000 (5)	0.007 (5)	0.010 (6)
C18	0.046 (7)	0.043 (7)	0.054 (7)	-0.005 (6)	0.007 (5)	0.005 (6)
C19	0.044 (7)	0.051 (7)	0.064 (7)	-0.001 (6)	0.012 (5)	-0.001 (6)
C20	0.047 (7)	0.057 (7)	0.071 (7)	0.007 (6)	0.013 (6)	-0.003 (6)
C21	0.043 (7)	0.056 (7)	0.076 (8)	0.006 (6)	0.008 (6)	0.006 (7)
C22	0.038 (6)	0.051 (7)	0.055 (6)	-0.001 (5)	0.000 (5)	0.005 (5)
C23	0.043 (7)	0.043 (6)	0.047 (6)	0.003 (5)	0.000 (5)	-0.004 (5)
C24	0.049 (9)	0.042 (7)	0.047 (7)	-0.006 (6)	0.009 (6)	0.002 (6)
C25	0.069 (9)	0.056 (8)	0.064 (7)	0.005 (7)	0.017 (6)	-0.003 (6)
C26	0.090 (11)	0.049 (7)	0.063 (7)	-0.002 (7)	0.001 (7)	-0.009 (6)
C27	0.082 (12)	0.066 (10)	0.060 (9)	0.019 (8)	0.008 (8)	-0.011 (8)
C28	0.064 (8)	0.079 (10)	0.064 (8)	0.010 (7)	0.020 (6)	-0.002 (7)
C29	0.054 (8)	0.065 (8)	0.059 (7)	-0.001 (6)	0.012 (6)	-0.002 (6)
C30	0.052 (7)	0.040 (6)	0.054 (6)	0.002 (5)	0.008 (5)	0.003 (5)
C31	0.075 (10)	0.062 (9)	0.066 (8)	0.010 (8)	0.002 (7)	0.019 (7)
C32	0.088 (11)	0.067 (9)	0.077 (9)	0.003 (8)	0.010 (7)	0.028 (7)
C33	0.093 (11)	0.052 (8)	0.081 (9)	0.002 (8)	0.019 (8)	0.013 (7)
C34	0.089 (11)	0.049 (8)	0.079 (9)	-0.006 (7)	0.015 (8)	0.003 (7)

supplementary materials

C35	0.084 (9)	0.050 (7)	0.063 (8)	-0.005 (7)	-0.022 (6)	0.009 (6)
C36	0.053 (9)	0.055 (9)	0.055 (9)	0.014 (7)	0.010 (7)	0.004 (7)
C37	0.072 (9)	0.078 (10)	0.062 (8)	-0.002 (7)	0.006 (7)	0.014 (7)
C38	0.086 (11)	0.092 (12)	0.059 (8)	0.011 (10)	0.008 (7)	0.018 (8)
C39	0.091 (12)	0.069 (11)	0.066 (9)	0.021 (9)	0.029 (9)	0.025 (8)
C40	0.100 (13)	0.058 (9)	0.082 (10)	0.008 (8)	0.033 (9)	0.008 (8)
C41	0.084 (10)	0.063 (9)	0.053 (7)	-0.001 (7)	0.016 (6)	0.008 (6)
C42	0.073 (10)	0.063 (9)	0.077 (8)	-0.001 (7)	0.002 (7)	0.005 (7)
C43	0.100 (12)	0.071 (10)	0.098 (11)	-0.005 (9)	0.002 (9)	0.004 (8)
C44	0.068 (9)	0.075 (9)	0.076 (8)	0.009 (7)	0.010 (7)	-0.004 (7)
C45	0.109 (14)	0.098 (13)	0.107 (12)	0.022 (11)	0.026 (12)	-0.022 (12)
C46	0.067 (9)	0.062 (9)	0.085 (9)	0.006 (7)	-0.003 (7)	-0.012 (8)
C47	0.092 (12)	0.092 (12)	0.078 (10)	0.017 (9)	-0.024 (8)	-0.001 (9)

Geometric parameters (Å, °)

Sn1—C36	2.154 (15)	C23—H23A	0.9700
Sn1—C24	2.157 (14)	C23—H23B	0.9700
Sn1—C30	2.180 (11)	C24—C25	1.403 (19)
Sn1—O4	2.227 (12)	C24—C29	1.412 (19)
Sn1—O1	2.314 (12)	C25—C26	1.410 (17)
N1—C44	1.507 (16)	C25—H25	0.9300
N1—C46	1.507 (16)	C26—C27	1.37 (2)
N1—C42	1.531 (16)	C26—H26	0.9300
N1—H1	0.9100	C27—C28	1.39 (2)
O1—C1	1.307 (15)	C27—H27	0.9300
O2—C1	1.248 (13)	C28—C29	1.398 (17)
O3—C4	1.371 (17)	C28—H28	0.9300
O3—H3	0.8200	C29—H29	0.9300
O4—C12	1.263 (15)	C30—C35	1.344 (17)
O5—C12	1.276 (14)	C30—C31	1.429 (17)
O6—C15	1.358 (12)	C31—C32	1.387 (19)
O6—H6	0.8200	C31—H31	0.9300
C1—C3	1.502 (15)	C32—C33	1.41 (2)
C2—C3	1.390 (15)	C32—H32	0.9300
C2—C7	1.413 (15)	C33—C34	1.37 (2)
C2—H2	0.9300	C33—H33	0.9300
C3—C4	1.44 (2)	C34—C35	1.382 (18)
C4—C5	1.40 (2)	C34—H34	0.9300
C5—C6	1.456 (16)	C35—H35	0.9300
C5—C23	1.509 (16)	C36—C41	1.38 (2)
C6—C7	1.406 (15)	C36—C37	1.401 (19)
C6—C11	1.431 (15)	C37—C38	1.398 (18)
C7—C8	1.442 (16)	C37—H37	0.9300
C8—C9	1.363 (16)	C38—C39	1.39 (2)
C8—H8	0.9300	C38—H38	0.9300
C9—C10	1.386 (17)	C39—C40	1.37 (2)
C9—H9	0.9300	C39—H39	0.9300
C10—C11	1.340 (17)	C40—C41	1.399 (18)

C10—H10	0.9300	C40—H40	0.9300
C11—H11	0.9300	C41—H41	0.9300
C12—C14	1.505 (16)	C42—C43	1.537 (19)
C13—C14	1.379 (19)	C42—H42A	0.9700
C13—C18	1.39 (2)	C42—H42B	0.9700
C13—H13	0.9300	C43—H43A	0.9600
C14—C15	1.420 (14)	C43—H43B	0.9600
C15—C16	1.400 (15)	C43—H43C	0.9600
C16—C17	1.453 (15)	C44—C45	1.49 (2)
C16—C23 ⁱ	1.520 (15)	C44—H44A	0.9700
C17—C22	1.397 (16)	C44—H44B	0.9700
C17—C18	1.450 (17)	C45—H45A	0.9600
C18—C19	1.408 (17)	C45—H45B	0.9600
C19—C20	1.364 (16)	C45—H45C	0.9600
C19—H19	0.9300	C46—C47	1.48 (2)
C20—C21	1.375 (16)	C46—H46A	0.9700
C20—H20	0.9300	C46—H46B	0.9700
C21—C22	1.379 (16)	C47—H47A	0.9600
C21—H21	0.9300	C47—H47B	0.9600
C22—H22	0.9300	C47—H47C	0.9600
C23—C16 ⁱⁱ	1.520 (15)		
C36—Sn1—C24	110.7 (4)	H23A—C23—H23B	106.9
C36—Sn1—C30	139.3 (5)	C25—C24—C29	117.8 (13)
C24—Sn1—C30	109.5 (5)	C25—C24—Sn1	123.5 (10)
C36—Sn1—O4	91.1 (5)	C29—C24—Sn1	118.3 (10)
C24—Sn1—O4	89.3 (5)	C24—C25—C26	120.3 (13)
C30—Sn1—O4	95.1 (4)	C24—C25—H25	119.8
C36—Sn1—O1	86.3 (5)	C26—C25—H25	119.8
C24—Sn1—O1	89.2 (5)	C27—C26—C25	120.6 (14)
C30—Sn1—O1	88.6 (4)	C27—C26—H26	119.7
O4—Sn1—O1	176.3 (4)	C25—C26—H26	119.7
C44—N1—C46	113.6 (10)	C26—C27—C28	120.5 (14)
C44—N1—C42	114.0 (10)	C26—C27—H27	119.7
C46—N1—C42	109.4 (10)	C28—C27—H27	119.7
C44—N1—H1	106.4	C27—C28—C29	119.7 (13)
C46—N1—H1	106.4	C27—C28—H28	120.2
C42—N1—H1	106.4	C29—C28—H28	120.2
C1—O1—Sn1	114.2 (8)	C28—C29—C24	121.0 (13)
C4—O3—H3	109.5	C28—C29—H29	119.5
C12—O4—Sn1	120.9 (9)	C24—C29—H29	119.5
C15—O6—H6	109.5	C35—C30—C31	117.8 (11)
O2—C1—O1	122.6 (11)	C35—C30—Sn1	123.9 (9)
O2—C1—C3	121.3 (10)	C31—C30—Sn1	118.3 (9)
O1—C1—C3	116.0 (10)	C32—C31—C30	118.1 (15)
C3—C2—C7	121.4 (10)	C32—C31—H31	121.0
C3—C2—H2	119.3	C30—C31—H31	121.0
C7—C2—H2	119.3	C31—C32—C33	122.3 (14)
C2—C3—C4	118.1 (11)	C31—C32—H32	118.9

supplementary materials

C2—C3—C1	118.5 (10)	C33—C32—H32	118.9
C4—C3—C1	123.3 (11)	C34—C33—C32	117.6 (13)
O3—C4—C5	118.3 (14)	C34—C33—H33	121.2
O3—C4—C3	119.1 (15)	C32—C33—H33	121.2
C5—C4—C3	122.6 (13)	C33—C34—C35	119.6 (14)
C4—C5—C6	117.7 (12)	C33—C34—H34	120.2
C4—C5—C23	120.1 (11)	C35—C34—H34	120.2
C6—C5—C23	122.1 (12)	C30—C35—C34	124.0 (13)
C7—C6—C11	117.1 (10)	C30—C35—H35	118.0
C7—C6—C5	119.9 (10)	C34—C35—H35	118.0
C11—C6—C5	123.0 (10)	C41—C36—C37	117.0 (14)
C6—C7—C2	120.3 (9)	C41—C36—Sn1	118.1 (11)
C6—C7—C8	119.8 (10)	C37—C36—Sn1	124.9 (12)
C2—C7—C8	119.8 (10)	C38—C37—C36	121.3 (15)
C9—C8—C7	120.0 (11)	C38—C37—H37	119.4
C9—C8—H8	120.0	C36—C37—H37	119.4
C7—C8—H8	120.0	C39—C38—C37	119.4 (14)
C8—C9—C10	119.3 (11)	C39—C38—H38	120.3
C8—C9—H9	120.3	C37—C38—H38	120.3
C10—C9—H9	120.3	C40—C39—C38	120.7 (14)
C11—C10—C9	122.5 (12)	C40—C39—H39	119.6
C11—C10—H10	118.8	C38—C39—H39	119.6
C9—C10—H10	118.8	C39—C40—C41	118.9 (15)
C10—C11—C6	121.0 (11)	C39—C40—H40	120.5
C10—C11—H11	119.5	C41—C40—H40	120.5
C6—C11—H11	119.5	C36—C41—C40	122.5 (14)
O4—C12—O5	123.6 (11)	C36—C41—H41	118.8
O4—C12—C14	118.4 (11)	C40—C41—H41	118.8
O5—C12—C14	118.0 (10)	N1—C42—C43	113.0 (11)
C14—C13—C18	124.6 (15)	N1—C42—H42A	109.0
C14—C13—H13	117.7	C43—C42—H42A	109.0
C18—C13—H13	117.7	N1—C42—H42B	109.0
C13—C14—C15	117.7 (12)	C43—C42—H42B	109.0
C13—C14—C12	121.1 (12)	H42A—C42—H42B	107.8
C15—C14—C12	121.1 (10)	C42—C43—H43A	109.5
O6—C15—C16	118.2 (9)	C42—C43—H43B	109.5
O6—C15—C14	120.0 (9)	H43A—C43—H43B	109.5
C16—C15—C14	121.8 (9)	C42—C43—H43C	109.5
C15—C16—C17	119.2 (10)	H43A—C43—H43C	109.5
C15—C16—C23 ⁱ	119.2 (9)	H43B—C43—H43C	109.5
C17—C16—C23 ⁱ	121.5 (10)	C45—C44—N1	114.9 (12)
C22—C17—C18	117.8 (10)	C45—C44—H44A	108.5
C22—C17—C16	123.4 (11)	N1—C44—H44A	108.5
C18—C17—C16	118.7 (11)	C45—C44—H44B	108.5
C13—C18—C19	124.0 (13)	N1—C44—H44B	108.5
C13—C18—C17	117.8 (13)	H44A—C44—H44B	107.5
C19—C18—C17	118.2 (11)	C44—C45—H45A	109.5
C20—C19—C18	122.3 (11)	C44—C45—H45B	109.5

C20—C19—H19	118.9	H45A—C45—H45B	109.5
C18—C19—H19	118.9	C44—C45—H45C	109.5
C19—C20—C21	118.9 (11)	H45A—C45—H45C	109.5
C19—C20—H20	120.6	H45B—C45—H45C	109.5
C21—C20—H20	120.6	C47—C46—N1	114.0 (12)
C20—C21—C22	122.1 (11)	C47—C46—H46A	108.8
C20—C21—H21	119.0	N1—C46—H46A	108.8
C22—C21—H21	119.0	C47—C46—H46B	108.8
C21—C22—C17	120.8 (11)	N1—C46—H46B	108.8
C21—C22—H22	119.6	H46A—C46—H46B	107.6
C17—C22—H22	119.6	C46—C47—H47A	109.5
C5—C23—C16 ⁱⁱ	120.3 (11)	C46—C47—H47B	109.5
C5—C23—H23A	107.3	H47A—C47—H47B	109.5
C16 ⁱⁱ —C23—H23A	107.3	C46—C47—H47C	109.5
C5—C23—H23B	107.3	H47A—C47—H47C	109.5
C16 ⁱⁱ —C23—H23B	107.3	H47B—C47—H47C	109.5

Symmetry codes: (i) $x+1, -y+1, z+1/2$; (ii) $x-1, -y+1, z-1/2$.

Fig. 1

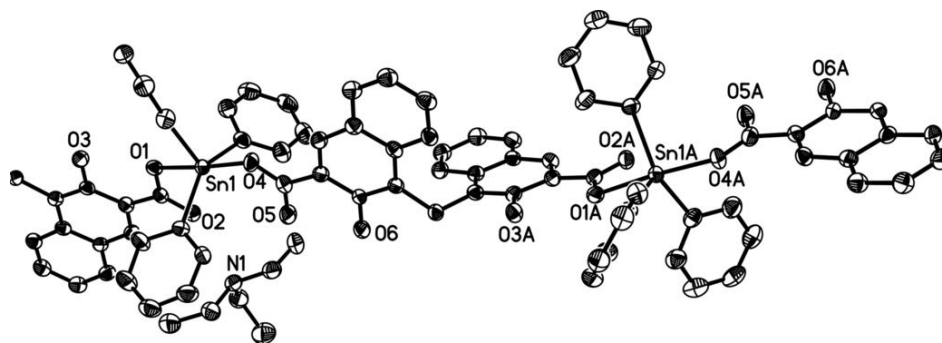


Fig. 2

